

origins would offer one possible explanation for the approximate constancy of screening factors found by Read⁶ for multiply-excited states.

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Infrared multiphoton dissociation of SF₆ in a molecular beam: Observation of F atoms by chemi-ionization detection

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Ever since SF₆ has been observed to dissociate in an isotopically selective manner following multiphoton absorption from an intense infrared radiation field,^{1,2} there has been considerable effort, both experimental and theoretical, to understand the mechanism responsible for this process.³ Of particular importance is the identification of the initial photofragments. From mass spectrometric and infrared studies Kompa and co-workers⁴ suggested that SF₆ initially decomposes into SF₄ + F₂, while from molecular beam studies Coggiola, Schulz, Lee, and Shen⁵ concluded that dissociation proceeds via the lower energy path, SF₅ + F. Further work by the same

group⁶ shows that SF₅ undergoes secondary absorption and decomposition to SF₄ + F. We have crossed a beam of SF₆ with the output of a high power pulsed CO₂ laser. Using a new chemi-ionization technique to distinguish between F₂ and F fragments, we report here results that corroborate the findings of Coggiola *et al.*, namely, only F atoms are formed in the infrared multiphoton dissociation of SF₆.

The experimental setup is shown in Fig. 1. An atomic samarium ($\geq 99.9\%$) beam, issuing from an electrically heated oven, intersects a SF₆ ($\geq 99.99\%$) beam in

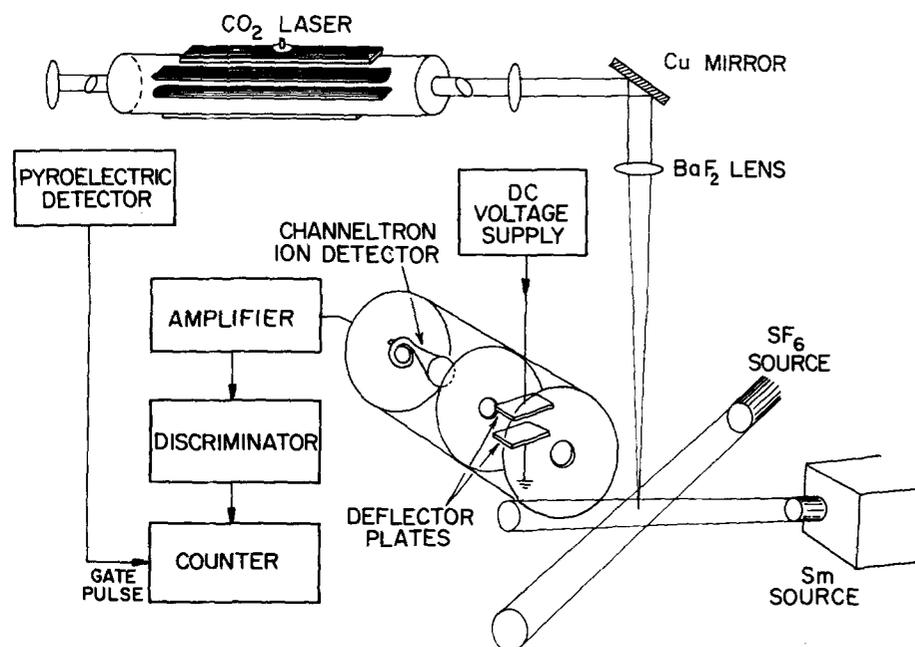


FIG. 1. Schematic of the experimental apparatus. The output of a 10.6 μ CO₂ laser is focused onto a molecular beam of SF₆. The photofragments react with an atomic Sm beam to form ions which are detected by a channeltron. Variation of the voltage on the deflector plates permits discrimination of electrons from negative ions.

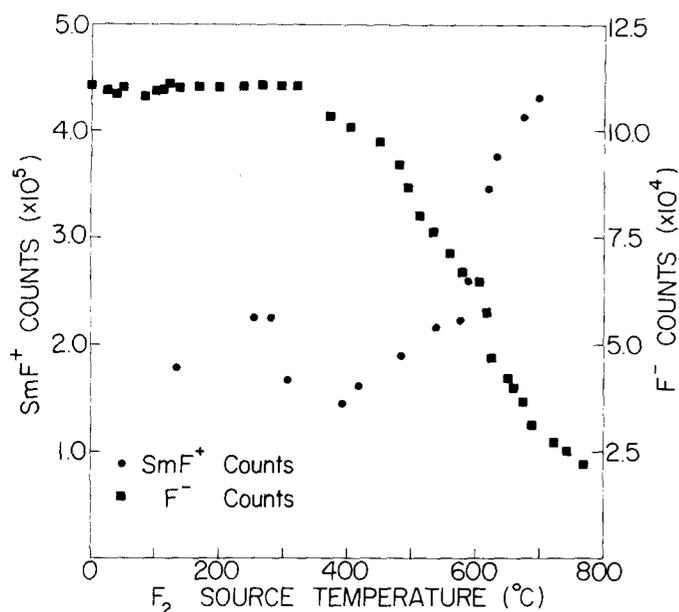


FIG. 2. The SmF^+ and F^- signals from a quadrupole mass spectrometer recorded as a function of the F_2 nozzle temperature. The increase in the SmF^+ signal concurrent with the decrease in the F^- signal at temperatures where the F_2 is significantly dissociated establishes the presence of Reaction (2).

an evacuated chamber (8×10^{-6} torr). The output of a transversely excited CO_2 laser (~ 1 J/p, 1 pps, several hundred nanosecond pulsewidth) operating on the 10.6μ $P(20)$ line is focused by a BaF_2 biconvex lens ($f/5$) into the intersection region of the Sm and SF_6 beams. A channeltron (charged particle multiplier) equipped with electric deflection plates (whose field is perpendicular to the path of the ions) is situated near the intersection region. The channeltron output is counted for a fixed period of time ($30 \mu\text{sec}$) so that only ions formed within

a short time after the beginning of the laser pulse are recorded. A signal from a pyroelectric detector viewing the laser output triggers the opening of the counter circuit.

In a separate experiment we have observed (Fig. 2) that the following thermal chemi-ionization reactions occur under crossed-beam conditions:



and



In Reactions (1) and (2) both positive and negative ions were identified using a quadrupole mass filter.⁷ We have also found that no ions are formed under the same conditions for reactions of Sm with SF_4 or SF_6 , even when the latter is heated to 670°C . Thus, by discriminating between F^- and e^- , Reactions (1) and (2) serve to determine the initial photofragment distribution following infrared multiphoton dissociation of SF_6 . This is readily accomplished by electrostatic deflection in conjunction with the earth's magnetic field, as illustrated in Fig. 3(a), since the application of a few volts to the deflector plates in front of the channeltron is sufficient to remove the lighter electrons but to leave the transmission of the heavier F^- ions almost unaffected.⁸

Figure 3(b) shows both the positive and negative charged particle counts observed following multiphoton dissociation of SF_6 . The pronounced falloff of the negative current with applied voltage is strong evidence that the negatively charged species are only electrons [compare Fig. 3(b) with 3(a)]. We conclude therefore that the initial photofragmentation of SF_6 involves the production of F atoms.

We thank S. Freund and J. Hudson for construction of

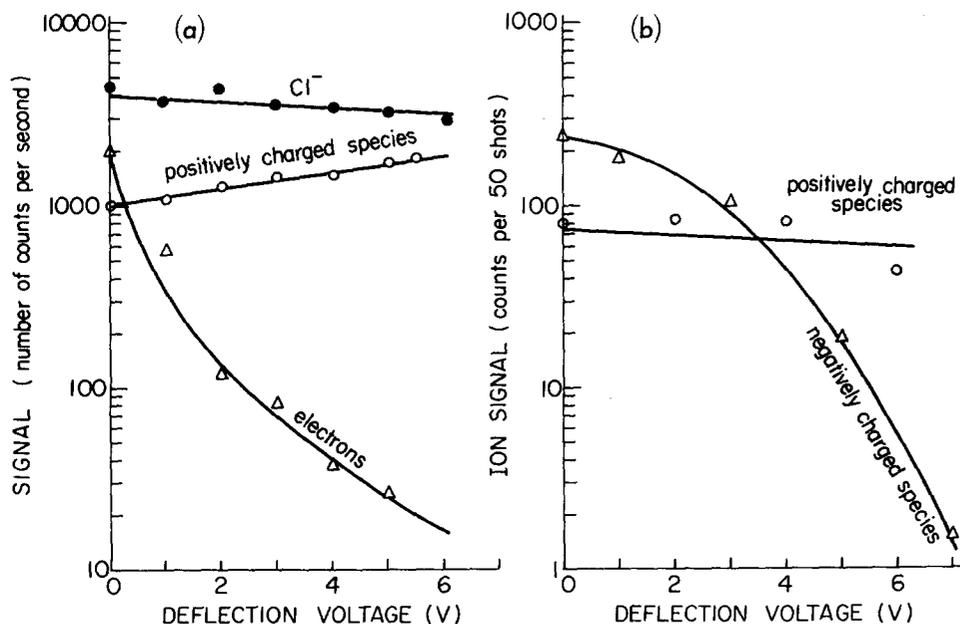


FIG. 3. Ion counts as a function of deflection plate voltage: (a) electrons from a heated tantalum filament in addition to Cl^- ions (generated by the chemi-ionization reaction $\text{Sm} + \text{Cl}_2 \rightarrow \text{SmCl}^+ + \text{Cl}^-$); and (b) chemi-ions from the reaction of Sm with photofragments following SF_6 multiphoton dissociation.

of the CO₂ laser, as well as M. Alexandre for assistance in the preliminary stages of these experiments. We are also grateful to W. C. Lineberger for helpful discussions. This work is supported by the Energy Research and Development Administration.

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⁸Crossed electric and magnetic fields of unequal magnitude are equivalent in a moving frame to a static magnetic field that is weaker. As the electric field is increased, the radius of curvature of the electrons varies from a value less than to one greater than the aperture diameter, whereas that of the heavier (positive or negative) ions is larger than the aperture diameter under all conditions. See J. D. Jackson, *Classical Electrodynamics* (Wiley, New York, 1962), pp. 412–414.

COMMENTS

Absolute coverage determination of CO on nickel crystal faces: Papers by Erley, Besocke, and Wagner [*J. Chem. Phys.* **66**, 5269 (1977)] and by Klier, Zettlemoyer, and Leidheiser [*J. Chem. Phys.* **52**, 589 (1970)]

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Erley, Besocke, and Wagner¹ recently made a critical comment that no remark was made by Klier, Zettlemoyer, and Leidheiser² on the backscatter coefficient of ¹⁴C radiation in their description of ¹⁴C tracer calibrations employed in the determination of CO coverages on nickel substrates, and suggested that an error of up to 50% surface coverage may have ensued from the neglect of backscattering. The purpose of this comment is to communicate that the backscatter corrections were in fact made and included in counter efficiency stated in the original paper by Klier, Zettlemoyer, and Leidheiser.² Although the backscatter correction was not explicitly stated, the data in the paper (Ref. 2) do include a backscatter factor (f_B) of 1.34 for the nickel substrate: the reported count rate (R) of 3140 cpm/nM of adsorbed ¹⁴C of specific activity (σ) 22.7 Ci/M at 9.3% counter efficiency (E) defines the backscatter coefficient at $f_B = (2 \times R)/(E \times \sigma) = 2 \times 3140 \times 10^9 / (0.093 \times 22.7 \times 2.22 \times 10^{12}) = 1.34$, where 1 Ci = 2.22×10^{12} dpm and the factor 2 stands for the ratio of 4π to 2π geometry. The backscatter coefficient of 1.34 was taken from Burt's work³ for ⁶⁰Co beta radiation (0.3 MeV) on Cu ($Z = 29$)

in the absence of data for ¹⁴C (0.16 MeV) on Ni ($Z = 28$). Burt's backscatter factors were chosen in preference to Glendenin and Solomon's,⁴ who give $f_B = 1.5$ for 0.17 MeV radiation on $Z = 28$, because in the latter work the backscatter was reported to vary with an insertion of a collodion film considered to have zero backscattering between the source (⁸⁵S) and the metal. The variations were attributed to surface roughness the effects of which may have not been totally removed by the presence of the collodion film.

A further support in favor of the calibration procedure used by Klier *et al.*² involving $f_B = 1.34$ comes from Tracy's⁵ independent AES measurements of CO saturation coverage on Ni (100) which agrees to within 1% with that of Klier *et al.* and from Taylor and Estrup's⁶ excellent agreement of LEED structures with CO coverages on Ni (110) determined from AES isotherms normalized to saturation coverage obtained from Ref. 2.

This Comment reiterates the already recorded agreement between results of papers Refs. 2, 5, and 6, concerning CO adsorption on the Ni (100) and Ni (110) faces.