

treme and prolonged environmental stress (39) and were likely unaffected. Many eukaryotic phyla (including red, green, and chromophytic algae) evolved before the late Neoproterozoic glaciations and also must have survived the environmental stress (40). However, a succession of snowball glaciations must have imposed an intense environmental filter, resulting in a series of genetic "bottleneck and flush" cycles (41), possibly leading to an initial metazoan radiation before the terminal glaciation (42) and an Ediacaran radiation in its aftermath (11).

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Photofragment Helicity Caused by Matter-Wave Interference from Multiple Dissociative States

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Isolated diatomic molecules of iodine monochloride (ICl) were photodissociated by a beam of linearly polarized light, and the resulting ground-state Cl atom photofragments were detected by a method that is sensitive to the handedness (helicity) of the electronic angular momentum. It was found that this helicity oscillates between "topspin" and "backspin" as a function of the wavelength of the dissociating light. The helicity originates solely from the (de Broglie) matter-wave interference of multiple dissociating pathways of the electronic excited states of ICl. These measurements can be related to the identity and to the detailed shapes of the dissociating pathways, thus demonstrating that it is possible to probe repulsive states by spectroscopic means.

The photodissociation of a diatomic molecule occurs, in the simplest case, as the breakup of an excited molecule on a single potential

energy surface (1). The molecule then dissociates under the influence of a force directed along the bond axis. The photodissociation

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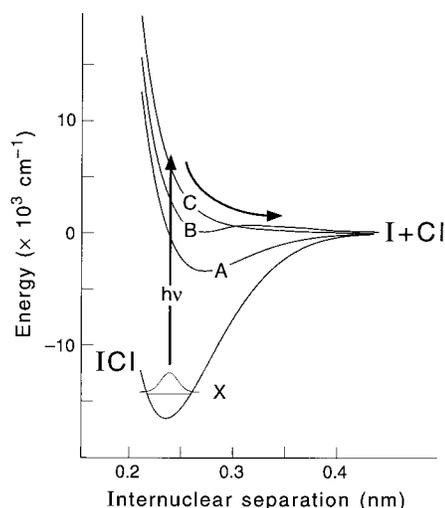


Fig. 1. Potential energy curves as a function of the internuclear separation for the ground electronic state (labeled X) and the three lowest excited electronic states (labeled A, B, and C) of ICl. At large internuclear separations, all three excited-state curves correlate to ground-state I and Cl atoms. The absorption of linearly polarized light in the range of 490 to 560 nm ($h\nu$ is the energy of a photon) causes ICl to dissociate to I and Cl atoms via the three dissociative pathways, indicated by a curved arrow. Interference between the $\Omega' = 0$ (B state) and $\Omega' = 1$ pathways (A and C states) produces I and Cl atoms with oriented angular momenta (with topspin or backspin).

products are often open-shell atoms with electronic angular momenta. In single-surface photodissociations with linearly polarized light, the symmetry of the dissociation process constrains the electronic angular momentum of the atomic photofragment to lack helicity, that is, “up” is equivalent to “down” and “left” is equivalent to “right.” In the more general case, the dissociative transition at a fixed photon energy can prepare more than one excited state, and the molecule can break apart on several potential energy surfaces simultaneously. Quantum interference between multiple electronic pathways in the dissociation has been analyzed theoretically (2–5), and it has been shown that it can lead to helicity in the angular momenta of the atomic fragments (6). We report an experimental observation of this effect in the photodissociation of ICl and demonstrate how interference effects in molecular dissociation cause helicity of the photofragment angular momenta, which vary in an oscillatory fashion with photolysis energy. This oscillation of electronic helicity depends sensitively on the shapes of the interfering potential energy curves. Just as interference in x-ray diffrac-

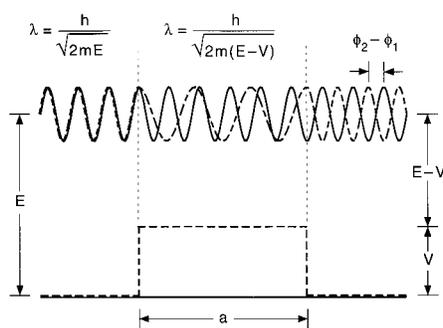


Fig. 2. Development of the asymptotic phase shift, $(\phi_2 - \phi_1)$, between two matter waves of energy E , one of which follows a flat potential (solid), and the other of which encounters an energy barrier of height V and width a (dashed). The de Broglie wavelength is a function of E , V , the mass, m , and Planck's constant, h . Note that the two matter waves have different wavelengths only above the barrier.

tion is used to deduce molecular structure, we demonstrate how interference in photodissociation can be used to deduce the shape of molecular excited states. The measurements presented here constitute a spectroscopic method for analyzing repulsive excited-state potentials and permit a rigorous test of the calculation of these potentials.

Electrons in atoms and molecules have two types of angular momentum: the intrinsic electronic spin and electronic orbital angular momentum. In general, the electrons that constitute a chemical bond pair up their spins such that the total spin of the molecule is zero. Similarly, the total electronic orbital angular momentum usually has no net projection along a bond axis. Therefore, the net projection of the total angular momentum along the bond axis, Ω , is zero for the ground state of most diatomic molecules. The absorption of a single photon can promote the transition of the diatomic molecule from the $^1\Sigma^+$ ($\Omega = 0$) ground state labeled X in Fig. 1, to excited electronic states with $\Omega' = 0$ or $\Omega' = 1$, known as parallel and perpendicular transitions, respectively. For example, ab initio calculations (7, 8) have shown that the absorption of green light from the $\Omega = 0$ ground state of ICl leads to two $\Omega' = 1$ excited states (the $^3\Pi_1$ and $^1\Pi_1$ states, labeled as A and C, respectively) through a perpendicular transition, and to one $\Omega' = 0$ excited state (the $^3\Pi_{0+}$ state, labeled as B) through a parallel transition (Fig. 1). Parallel transitions produce photofragments with velocities preferentially parallel to the electric vector, ϵ_{phot} , of the linearly polarized photolysis light (in a $\cos^2 \theta$ distribution, where θ is the angle between ϵ_{phot} and the recoil direction), whereas perpendicular transitions produce photofragments with velocities preferentially perpendicular to the electric vector of the linearly polarized

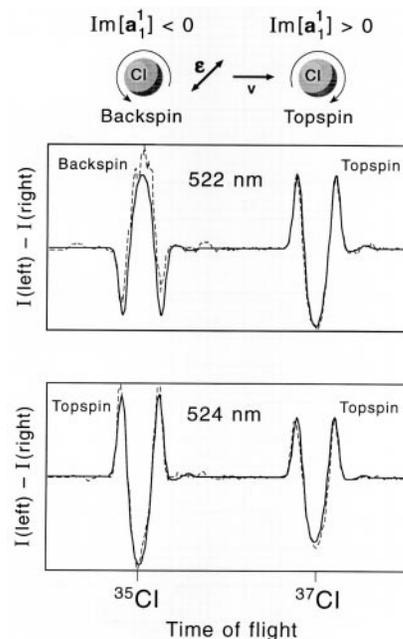


Fig. 3. Experimental time-of-flight difference profiles for ^{35}Cl and ^{37}Cl atoms [left minus right circularly polarized light [$I(\text{left})$ and $I(\text{right})$, respectively]]. The profiles are centered at arrival times of 4.80 and 4.94 μs , respectively. These difference profiles have a characteristic shape, and their size is proportional to $\text{Im}[a_1^1]$ (photofragment helicity). The motions of topspin and backspin are defined with respect to the Cl-atom velocity, \mathbf{v} , and the electric vector of the dissociating light, ϵ_{phot} , as shown. A positive $\text{Im}[a_1^1]$ value describes angular momenta pointing preferentially into the page, corresponding to a motion with topspin (with respect to the vectors ϵ_{phot} and \mathbf{v}), whereas a negative $\text{Im}[a_1^1]$ value describes angular momenta pointing preferentially out of the page, corresponding to a motion with backspin.

photolysis light (in a $\sin^2 \theta$ distribution) (9). Measurements of the spatial distribution of the photofragment velocities as a function of the dissociating wavelength show that at 490 nm the photodissociation occurs almost entirely through the $\Omega' = 0$ state; at 560 nm the photodissociation occurs almost entirely through one or both of the $\Omega' = 1$ states; and at intermediate wavelengths a mixture of $\Omega' = 0$ and $\Omega' = 1$ states are accessed (10). Measurements of the spatial distribution of the photofragments do not allow the determination of the relative contribution of the two $\Omega' = 1$ states.

The origin of photofragment helicity can be explained in classical terms by considering the motion of the charge distribution. For a parallel (or perpendicular) transition, the absorption of a linearly polarized photon causes the charge distribution to oscillate parallel (or perpendicular) to the internuclear axis. Upon separation of the diatomic molecule into its two atomic part-

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ners, no orientation (helicity) of the electronic angular momentum of the atomic photofragments can occur in either case. In a “mixed transition” involving excitation of both parallel and perpendicular transitions, the electron charge cloud initially oscillates along a line that makes an angle with respect to the internuclear axis. This motion can be decomposed into components parallel and perpendicular to the internuclear axis, oscillating in phase. As the nuclei separate, the frequencies of oscillation of these two components can differ so that they are no longer oscillating in phase. The frequency difference is directly related to the energy difference between the interfering dissociating pathways. At large nuclear separations the energy difference vanishes and the frequencies of oscillation become equal again, but with a possible net phase shift (Fig. 2). Thus, the phase shift causes the electron charge cloud to develop elliptical motion (a phase shift of 90° results in pure circular motion), and the resultant

electronic angular momenta of the atomic photofragments have helicity.

This production of oriented (circularly polarized) matter is analogous to the production of circularly polarized light when linearly polarized light passes through a birefringent crystal, such as a quarter-wave plate (11). In the photolysis of ICl, the linearly polarized excitation light (and its associated angular momentum) is separated into components that excite parallel and perpendicular transitions. If a phase difference arises between these components as a result of the energetic differences of the two paths followed by the separating nuclei, then the resulting Cl atoms can have oriented (circularly polarized) angular momenta. The introduction of a phase difference, $(\varphi_2 - \varphi_1)$, between two states separated by a constant potential energy, ΔE , and interacting for a time, t , is well studied in quantum-beat spectroscopy (12) and is given by

$$(\varphi_2 - \varphi_1) = \frac{\Delta E t}{\hbar} \quad (1)$$

where \hbar is Planck’s constant divided by 2π . The situation during photodissociation is very similar, except that the instantaneous potential energy difference between the two paths, ΔE , varies as a function of internuclear separation, and an appropriate time integration must be performed to determine the net phase difference.

An equivalent, and more visual, way of calculating the phase shift in the de Broglie waves associated with the nuclear motion along the two paths (Fig. 2). This method is equivalent because the difference in potential energy between the two paths at any instant is equal to the difference in kinetic energy of the atoms at the same instant. The de Broglie wavelength of the chlorine atoms, $\lambda(r)$, as a function of internuclear separation, r , is given by

$$\lambda(r) = \frac{h}{\sqrt{2m[E - V(r)]}} \quad (2)$$

where E is the total energy, $V(r)$ is the potential energy as a function of internuclear separation, m is the mass of the chlorine atom, and h is Planck’s constant. Different energetic pathways can support different numbers of de Broglie wavelengths (Fig. 2), so that a phase difference, $(\varphi_2 - \varphi_1)$, can be introduced into the two wave packets. The degree of helicity is proportional to $\sin(\varphi_2 - \varphi_1)$. Therefore, the determination of the photofragment helicity provides a direct measurement of the phase difference of de Broglie matter waves associated with the multiple dissociating pathways.

Our experimental approach is similar to that reported in (13). Briefly, ICl molecules from a pulsed supersonic beam expansion are photolyzed with linearly polarized light in the

range of 490 to 560 nm. The resulting ground-state Cl atom photofragments are ionized through absorption of three circularly polarized photons at 235 nm, to produce Cl^+ ions. The absorption of circularly polarized probe light is sensitive to the helicity of the angular momentum of the Cl atoms. The Cl^+ ions are detected with a time-of-flight mass spectrometer, and the width of the time-of-flight profiles (Fig. 3) is sensitive to the recoil velocity of the Cl atoms (13). The gross orientation of the electronic angular momenta of the Cl atoms can be described by the single dimensionless parameter $Im[\mathbf{a}_1^+]$ (2, 13–15), that in this case lies between $-\sqrt{3}/10$ and $\sqrt{3}/10$. This parameter describes a preference for an orientation of the angular momenta perpendicular to the plane defined by the photolysis laser polarization vector, \mathbf{e}_{phot} , and the Cl atom velocity, \mathbf{v} (Fig. 3).

The time-of-flight signal differences for left and right circularly polarized probe light at photolysis wavelengths of 522 and 524 nm are shown in Fig. 3. At a photolysis wavelength of 524 nm, both Cl isotopes have topspin, whereas at 522 nm, the ^{35}Cl photofragments have backspin and the ^{37}Cl photofragments have topspin (16). Similar measurements were performed in the 490- to 560-nm region, allowing the determination of the photofragment helicity, $Im[\mathbf{a}_1^+]$, as a function of photolysis wavelength for both Cl isotopes (Fig. 4A). In this wavelength region, the photolysis of ICl yields oriented Cl atoms that oscillate rapidly between topspin and backspin as a function of photolysis wavelength. This interference pattern represents a direct measurement of the phase difference of de Broglie wavelengths from the matter waves associated with multiple dissociating paths. The mass effect evident from the difference between the curves for ^{35}Cl and ^{37}Cl (Fig. 4A) arises from the mass dependence of the de Broglie wavelength shown in Eq. 2. Photolysis wavelength-dependent oscillations caused by coherence effects are discussed in the work of Siebbeles and Beswick (4). Note that the experimental oscillation pattern has an envelope that decays at both long and short wavelengths. The magnitude of the interference is greatest when the contributions from the $\Omega' = 0$ and $\Omega' = 1$ surfaces are equal, but tends toward zero when the contribution from either the $\Omega' = 0$ or $\Omega' = 1$ surface dominates. Spatial photofragment distribution measurements (10) have shown that the contributions from the $\Omega' = 0$ and $\Omega' = 1$ surfaces are equal at about 525 nm and diverge at both shorter and longer wavelengths. This behavior predicts an envelope for the oscillation of the orientation that peaks at 525 nm, in agreement with that observed. However, the spatial distribution measurements are not accurate enough to pre-

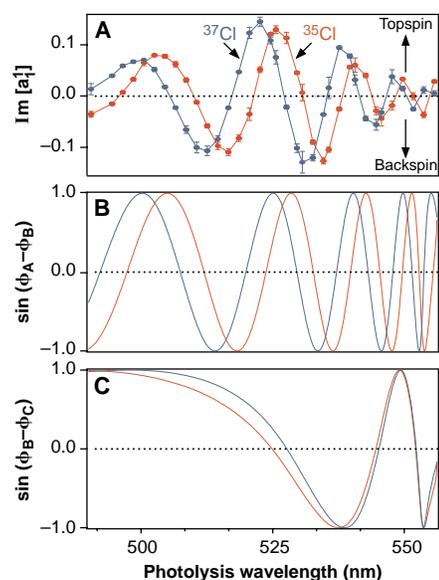


Fig. 4. (A) Experimental measurement of the degree of helicity, $Im[\mathbf{a}_1^+]$, of ^{35}Cl and ^{37}Cl from the photolysis of ICl as a function of the wavelength of the dissociating light. The magnitudes of the $Im[\mathbf{a}_1^+]$ parameter have been corrected for the effects of hyperfine depolarization (24, 25), and the error bars are statistical (1σ) resulting from replicate measurements. The angular momenta of the Cl atoms oscillate between topspin and backspin as a function of wavelength. (B) Calculated de Broglie phase shift $\sin(\varphi_A - \varphi_B)$ as a function of photolysis wavelength for dissociating surfaces A and B, shown in Fig. 1. (C) Calculated de Broglie phase shift $\sin(\varphi_B - \varphi_C)$ as a function of photolysis wavelength for dissociating surfaces B and C, shown in Fig. 1. Comparison to experiment shows clearly that the oscillations are matched well by the predicted interference between surfaces A and B, and very poorly by that between B and C.

dict the precise shape of the envelope, so we cannot rule out other factors contributing to the envelope shape.

We have solved numerically the one-dimensional Schrödinger equation for surfaces A, B, and C as a function of the dissociation energy to determine the energy-dependent phase differences, $(\varphi_A - \varphi_B)$ and $(\varphi_B - \varphi_C)$, in analogy to the simple example shown in Fig. 2. The photofragment orientation, $Im[\mathbf{a}_1^+]$, is linearly proportional to $\sin(\varphi_A - \varphi_B)$ and $\sin(\varphi_B - \varphi_C)$. Comparison of the photodissociation-wavelength dependence of these functions (Fig. 4, B and C) to experiment shows that the interference pattern is caused predominantly from dissociation via surfaces A and B. Overall, we conclude that at a photolysis wavelength of 490 nm, ICl dissociates predominantly via the B state; at 560 nm, ICl dissociates predominantly via the A state; and at intermediate wavelengths, the dissociation proceeds coherently through both states. The oscillations of $\sin(\varphi_A - \varphi_B)$ shown in Fig. 4B match very well the experimental oscillations in Fig. 4A, in contrast to the calculations shown in Fig. 4C. We conclude that the C state does not significantly participate in the photodissociation. The match at short wavelengths between Figs. 4A and 4B indicates that the calculated surfaces A and B are quite accurate; however, at longer wavelengths the calculated oscillations of $\sin(\varphi_A - \varphi_B)$ are too rapid. We suggest that this discrepancy may indicate that the shapes of the surfaces A and B need to be slightly improved.

To our knowledge, this study represents the first observation that the photolysis of a molecule with linearly polarized light causes the production of oriented photofragments. Previously, oriented photofragments have been observed using photolysis with circularly polarized light (17–22), but such orientation does not necessarily arise from matter-wave interference. This technique is not limited to diatomic molecules (23) and can be used to study excited states and dissociative processes with unprecedented sensitivity. Conventional spectroscopy is sensitive to the shape and nature of bound electronic states. In contrast, the interference pattern of the photofragment orientation is sensitive to the shape and nature of dissociating states. This technique can be used to investigate the unimolecular decomposition of chemical systems involving multiple electronic states by coherently preparing more than one electronic state of different symmetry. As such, the technique may become an important probe of wave-packet dynamics on multiple dissociative surfaces.

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 6. In the celebrated Young's two-slit experiment, the interference of light passing through two slits causes a bright and dark fringe pattern to appear on a screen, the intensity of which varies as $\cos\Delta\phi$ (where $\Delta\phi$ is the phase difference between the two paths). If instead, the light used is initially linearly polarized, and crossed polarizers are placed in front of the two slits (at 45° to the linear polarization so that both slits transmit equally), then the light on the screen no longer shows an intensity fringe pattern, but shows instead a polarization fringe pattern. This polarization pattern can be separated into components with linearly polarized variations (proportional to $\cos\Delta\phi$) and circularly polarized variations (proportional to $\sin\Delta\phi$). The analog of this special two-slit experiment in molecular photolysis is that in which the interference of matter waves from potential surfaces of different symmetry causes angular momentum polarization of the photofragments [for discussions and examples of $\cos\Delta\phi$ variations in the alignment of photofragments, see E. Flemming, O. Wilhelm, H. Schmoranz, M. Glass-Maujean, *J. Chem. Phys.* **103**, 4090 (1995) and (4, 5, 25)]. The experiment described in the present report is an example of the $\sin\Delta\phi$ variations in photofragment helicity.
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Reversal of Phencyclidine Effects by a Group II Metabotropic Glutamate Receptor Agonist in Rats

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Glutamatergic abnormalities have been associated with several psychiatric disorders, including schizophrenia and addiction. Group II metabotropic glutamate receptors were targeted to normalize glutamatergic disruptions associated with an animal model of schizophrenia, the phencyclidine model. An agonist of this group of receptors, at a dose that was without effects on spontaneous activity and cortic limbic dopamine neurotransmission, attenuated the disruptive effects of phencyclidine on working memory, stereotypy, locomotion, and cortical glutamate efflux. This behavioral reversal occurred in spite of sustained dopamine hyperactivity. Thus, targeting this group of receptors may present a nondopaminergic therapeutic strategy for treatment of psychiatric disorders.

Several lines of evidence suggest that glutamatergic mechanisms contribute to the pathophysiology of schizophrenia (1–3).

For example, phencyclidine (PCP) and other antagonists of *N*-methyl-D-aspartate (NMDA) receptors have psychotomimetic



Photofragment Helicity Caused by Matter-Wave Interference from Multiple Dissociative States

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