

## Correction to “Control of fossil-fuel particulate black carbon and organic matter, possibly the most effective method of slowing global warming”

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[1] This document describes two updates and a correction that affect two figures (Figures 1 and 14) in “Control of fossil-fuel particulate black carbon and organic matter, possibly the most effective method of slowing global warming” by Mark Z. Jacobson (*Journal of Geophysical Research*, 107(D19), 4410, doi:10.1029/2001JD001376, 2002). The modifications have no effect on the numerical simulations in the paper, only on the postsimulation analysis. The changes include the following: (1) The overall lifetime of CO<sub>2</sub> is updated to range from 30 to 95 years instead of 50 to 200 years, (2) the assumption that the anthropogenic emission rate of CO<sub>2</sub> is in equilibrium with its atmospheric mixing ratio is corrected, and (3) data for high-mileage vehicles available in the U.S. are used to update the range of mileage differences (15–30% better for diesel) in comparison with one difference previously (30% better mileage for diesel). The modifications do not change the main conclusions in J2002, namely, (1) “any emission reduction of fossil-fuel particulate BC plus associated OM may slow global warming more than may any emission reduction of CO<sub>2</sub> or CH<sub>4</sub> for a specific period,” and (2) diesel cars emitting continuously under the most recent U.S. and E.U. particulate standards (0.08 g/mi; 0.05 g/km) may warm climate per distance driven over the next 100+ years more than equivalent gasoline cars. Toughening vehicle particulate emission standards by a factor of 8 (0.01 g/mi; 0.006 g/km) does not change this conclusion, although it shortens the period over which diesel cars warm to 13–54 years,” except as follows: for conclusion 1, the period in Figure 1 of J2002 during which eliminating all fossil-fuel black carbon plus organic matter (f.f. BC + OM) has an advantage over all anthropogenic CO<sub>2</sub> decreases from 25–100 years to about 11–13 years and for conclusion 2 the period in Figure 14 of J2002 during which gasoline vehicles may have an advantage broadens from 13 to 54 years to 10 to >100 years. On the basis of the revised analysis, the ratio of the 100-year climate response per unit mass emission of f.f. BC + OM relative to that of CO<sub>2</sub>-C is estimated to be about 90–190.

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### 1. Lifetime of CO<sub>2</sub>

[2] In the work of Jacobson [2002, hereinafter referred to as J2002], it was assumed that the atmospheric lifetime of CO<sub>2</sub> against all loss processes combined was between 50 and 200 years. This range is commonly used in the literature. However, the upper lifetime does not appear to be physical, even within the range of reasonable uncertainty, and the lower lifetime appears to be too high to explain the rate of change of the observed mixing ratio of CO<sub>2</sub>.

[3] The data-constrained overall lifetime of CO<sub>2</sub> can be estimated as follows. First, the rate of change of the mixing

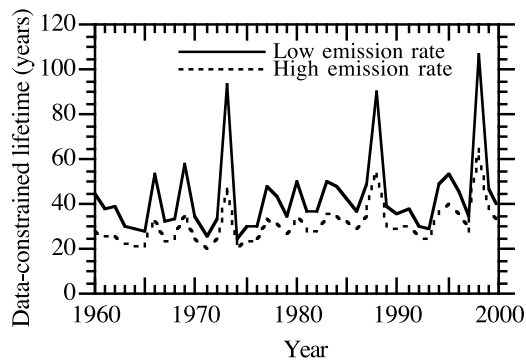
ratio ( $\chi$ , ppmv) of a well-mixed gas whose only source is emission is

$$\frac{d\chi(t)}{dt} = E - \frac{\chi(t)}{\tau}, \quad (1)$$

where  $E$  is the emission rate (ppmv/yr) and  $\tau$  is the overall  $e$ -folding lifetime (years) of the gas. Rearranging equation (1) gives the lifetime as

$$\tau = \frac{\chi(t)}{E - \frac{d\chi(t)}{dt}} \quad (2)$$

[e.g., Gaffin *et al.*, 1995]. Here, it is assumed that  $\chi(t)$  is the anthropogenic mixing ratio of CO<sub>2</sub> (the difference between the current mixing ratio and that during preindustrial times)



**Figure 1.** Data-constrained overall lifetime of CO<sub>2</sub> versus time calculated from equation (2) using yearly ambient CO<sub>2</sub> mixing ratio data from <http://cdiac.esd.ornl.gov/ftp/maunaloa-co2/maunaloa.co2>, yearly fossil-fuel CO<sub>2</sub> emission data from *Marland et al.* [2003], and biomass-burning emission rates ranging from 1500 to 2700 Tg-CO<sub>2</sub>-C/yr [Jacobson, 2004a]. The low and high emission rate curves in the figure represent the sum of the yearly fossil-fuel emission rate plus the fixed low or high biomass-burning emission rate. The 40-year (1960–2000) low- and high-emission rate mean data-constrained lifetimes are 43.0 and 30.6 years, respectively.

and  $E$  is the anthropogenic emission rate. These assumptions require the further assumption that the preindustrial mixing ratio ( $\chi_p(t) = 275$  ppmv in 1750) of CO<sub>2</sub> is in equilibrium with its natural emission rate,  $E_p$ . In other words,  $\chi_p(t) = \tau E_p$ , which is obtained by setting the derivative in equation (1) to zero.

[4] In the year 2000 ( $t = 0$ ), the overall mixing ratio of CO<sub>2</sub> was approximately 370 ppmv (available at <http://cdiac.esd.ornl.gov/ftp/maunaloa-co2/maunaloa.co2>), so the anthropogenic portion was about  $\chi(0) = 95$  ppmv ( $=370 - 275$  ppmv). From 1995 to 2000, the rate of change of the mixing ratio was about  $d\chi(0)/dt = 1.8$  ppmv/yr (available at <http://cdiac.esd.ornl.gov/ftp/maunaloa-co2/maunaloa.co2>). The global fossil-fuel emission rate of CO<sub>2</sub> in 2000 (and from 1995 to 2000) was near 6600 Tg-CO<sub>2</sub>-C/yr [Marland et al., 2003]. An estimated range of the anthropogenic portion of the outdoor biomass-burning emission rate is 1500–2700 Tg-CO<sub>2</sub>-C/yr [Jacobson, 2004a]. Thus the total global anthropogenic emission of CO<sub>2</sub> in 2000 may have ranged from 8100 to 9300 Tg-CO<sub>2</sub>-C/yr. With  $1.095602 \times 10^{44}$  air molecules in the global atmosphere (column abundance of air of  $2.14797 \times 10^{25}$  molec. cm<sup>-2</sup> and an area of the Earth of  $5.10064 \times 10^{18}$  cm<sup>2</sup>), this translates to a globally averaged emission rate of  $E = 3.7074 - 4.2566$  ppmv/yr (2184.82 Tg-CO<sub>2</sub>-C/yr = 1 ppmv/yr). Substituting the numbers above into equation (2) gives an estimated data-constrained lifetime of CO<sub>2</sub> for the year 2000 of 39–45 years.

[5] Figure 1 shows the data-constrained lifetime of CO<sub>2</sub> for 1960–2000, calculated using the methodology described. The lifetime ranged from 20 to 100 years, with an average between 30.6 and 43 years. Gaffin et al. [1995] performed a similar calculation with slightly different assumptions (preindustrial mixing ratio of 280 instead of 275 ppmv, a single biomass-burning emission rate, and for

the years 1959–1989) and found a mean lifetime on the order of 30 years. In no case in Figure 1 did the data-constrained lifetime approach 200 years. On the basis of Figure 1 and uncertainties associated with it, it is assumed here that the lifetime of CO<sub>2</sub> ranges from 30 to 95 years, although a more likely upper limit may be 50 or 60 years.

## 2. CO<sub>2</sub> Emissions Were No Longer Assumed To Be in Equilibrium

[6] The second update relates to the two CO<sub>2</sub> curves in Figure 1 of J2002. Each curve shows the estimated time-dependent temperature change due eliminating anthropogenic emission of CO<sub>2</sub> at a different assumed overall lifetime of CO<sub>2</sub> (50 or 200 years). The curves were obtained by running global climate response calculations at current and preindustrial mixing ratios of CO<sub>2</sub>, then scaling the resulting “equilibrium” temperature difference over time proportionally to the change in CO<sub>2</sub> mixing ratio over time. The CO<sub>2</sub> mixing ratio was assumed to be in equilibrium with its emission rate. Whereas the equilibrium assumption would hold under the current CO<sub>2</sub> emission rate if CO<sub>2</sub>’s lifetime were shorter (e.g.,  $\sim 25$  years or less) than it currently is or if CO<sub>2</sub>’s anthropogenic emission rate were lower than it currently is, this assumption is not valid under the current data-constrained lifetime or anthropogenic emission rate of CO<sub>2</sub>. Here, this assumption is corrected.

[7] Integrating equation (1) gives the analytical solution to the change in CO<sub>2</sub> mixing ratio over time as

$$\chi(t) = \chi(0)e^{-t/\tau} + \tau E(1 - e^{-t/\tau}). \quad (3)$$

Figure 2 here shows the time-dependent mixing ratio of CO<sub>2</sub> as a function of CO<sub>2</sub> lifetime for two respective emission rates from equation (3). In each case, an “equilibrium lifetime” exists (25.63 years and 22.32 years for the low and high emission rates, respectively), which is the lifetime at which the mixing ratio of CO<sub>2</sub> is always in equilibrium with a given emission rate (in other words, CO<sub>2</sub>’s mixing ratio is constant over time when the emission rate is constant). This equilibrium lifetime is  $\tau = \chi(0)/E$ , derived by setting  $\chi(t) = \chi(0)$  and solving for  $\tau$  in equation (3). It can also be derived by setting  $d\chi(t)/dt = 0$  in equation (1).

[8] The difference in the time-dependent mixing ratio when anthropogenic CO<sub>2</sub> emission is absent versus present is

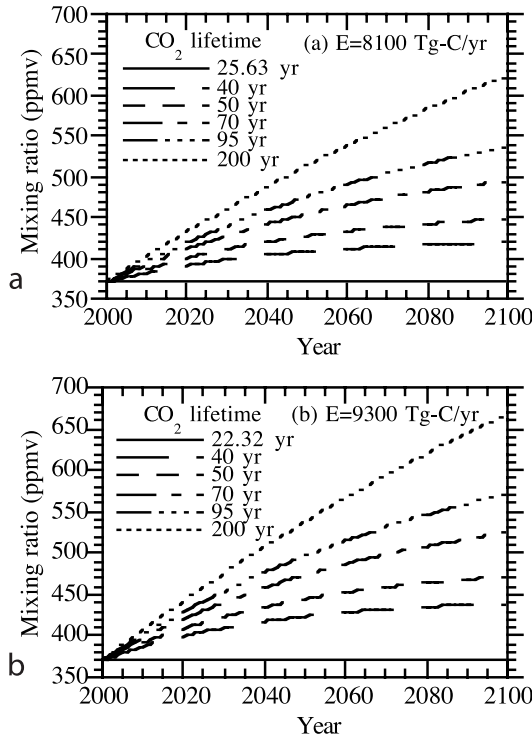
$$\Delta\chi(t) = [\chi(t)]_{\text{noemis}} - [\chi(t)]_{\text{w/emis}} = -\tau E(1 - e^{-t/\tau}), \quad (4)$$

where

$$\begin{aligned} [\chi(t)]_{\text{w/emis}} &= \chi(0)e^{-t/\tau} + \tau E(1 - e^{-t/\tau}) \\ [\chi(t)]_{\text{noemis}} &= \chi(0)e^{-t/\tau}, \end{aligned} \quad (5)$$

are obtained from equation (3) when  $E \neq 0$  and  $E = 0$ , respectively.

[9] J2002 assumed that when CO<sub>2</sub> was emitted, its emission rate was in equilibrium with its ambient mixing



**Figure 2.** Time-dependent mixing ratio of CO<sub>2</sub> versus year as a function of CO<sub>2</sub> lifetime for two constant emission rates. From equation (3) using 2184.82 Tg-CO<sub>2</sub>-C/yr = 1 ppmv/yr and  $\chi(0) = 95$  ppmv.

ratio ( $\tau = \chi(0)/E$ ). Substituting  $\tau E = \chi(0)$  into equation (4) gives

$$\Delta\chi(t) = -\chi(0)\left(1 - e^{-t/\tau}\right), \quad (6)$$

which was the mixing-ratio expression used to generate the CO<sub>2</sub> temperature-difference curves in Figure 1 of J2002.

[10] The equilibrium assumption is always correct when either (1) CO<sub>2</sub>'s lifetime equals its equilibrium lifetime ( $\tau = \tau_{eq} = \chi(0)/E$ , where  $E$  is the actual emission rate) for any time  $t$ , (2) CO<sub>2</sub>'s emission rate is constant for a sufficiently long period ( $t \gg \tau$  in equation (4)), or (3) CO<sub>2</sub>'s emission rate equals its equilibrium emission rate ( $E = E_{eq} = \chi(0)/\tau$ , where  $\tau$  is the actual lifetime).

[11] For example, when CO<sub>2</sub>'s actual emission rate is 9300 Tg-C/yr, Figure 2b shows that the equilibrium assumption is correct (1) for any  $t$  when CO<sub>2</sub>'s actual lifetime equals its equilibrium lifetime,  $\tau_{e0} = 22.3$  years or (2) for all lifetimes when  $t \gg \tau$ . Alternatively, the equilibrium assumption is correct (3) at an actual CO<sub>2</sub> lifetime of 31 years (Figure 1, lower curve) if CO<sub>2</sub>'s emission rate decreases to the equilibrium emission rate of  $E_{eq} = 6695$  Tg-CO<sub>2</sub>-C/yr.

[12] Figure 2, however, shows that under the current estimated range of CO<sub>2</sub> emission (8100–9300 Tg-C/yr) and under the current estimated range of CO<sub>2</sub> lifetime (30–95 years, from Figure 1), the mixing ratio of CO<sub>2</sub> is not in equilibrium with its emission rate. As such, the CO<sub>2</sub> mixing ratio will increase with time at a constant emission rate. For example, for average estimated CO<sub>2</sub> lifetimes of 31 years and 43 years from Figure 1 and a current emission rate of about 9300 and 8100 Tg-C/yr resulting in those

respective lifetimes, the anthropogenic CO<sub>2</sub> mixing ratio will increase from 95 ppmv to 132 and 159 ppmv, respectively, over the next 100 years. Similarly, for every 1000 Tg-C/yr increase in the emission rate, the mixing ratio should increase by another 14–20 ppmv.

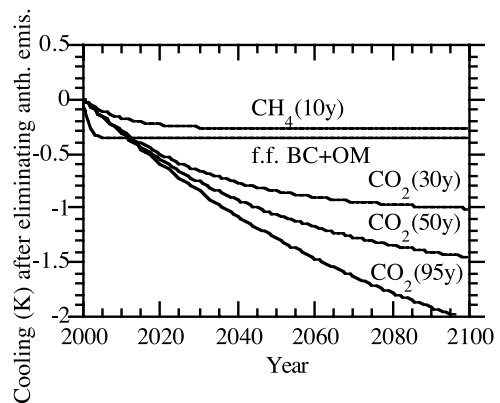
[13] To revise Figure 1 of J2002 with the information above, it is necessary to recalculate the estimated temperature change over time due to the time-dependent mixing ratio change from equation (4). Climate-response simulations from J2002 showed that the temperature change per unit mixing ratio of CO<sub>2</sub> differed upon a decrease (eliminating all anthropogenic emission) of CO<sub>2</sub> versus an increase (doubling) of CO<sub>2</sub>. Eliminating the anthropogenic mixing ratio of CO<sub>2</sub> ( $\Delta\chi_{eq,dec} = -95$  ppmv) resulted in an equilibrium temperature decrease of  $\Delta T_{eq,dec} = -0.9$  K whereas doubling CO<sub>2</sub> ( $\Delta\chi_{eq,inc} = 370$  ppmv) resulted in an equilibrium temperature increase of  $\Delta T_{eq,inc} = 3.2$  K. The reason for the different climate response per unit mixing ratio is that the response is a function of the mixing ratio itself and the feedbacks associated with it.

[14] The time-dependent temperature change accounting for the different climate responses upon a decrease or increase in mixing ratio is

$$\begin{aligned} \Delta T(t) &= \left\{ [\chi(t)]_{noemis} - \chi(0) \right\} \frac{\Delta T_{eq,dec}}{\Delta\chi_{eq,dec}} \\ &+ \left\{ \chi(0) - [\chi(t)]_{w/emis} \right\} \frac{\Delta T_{eq,inc}}{\Delta\chi_{eq,inc}} \\ &= \chi(0) \left( e^{-t/\tau} - 1 \right) \frac{\Delta T_{eq,dec}}{\Delta\chi_{eq,dec}} \\ &+ (\chi(0) - \tau E) \left( 1 - e^{-t/\tau} \right) \frac{\Delta T_{eq,inc}}{\Delta\chi_{eq,inc}} \end{aligned} \quad (7)$$

where the second expression was obtained by substituting equation (5) into the first. This equation differs from that used in J2002 only in that J2002 assumed  $\tau E = \chi(0)$ , resulting in  $\Delta T(t) = \chi(0)(e^{-t/\tau} - 1)\Delta T_{eq,dec}/\Delta\chi_{eq,dec}$ .

[15] Figure 3 shows modified time-dependent temperature-change curves when equation (7) is used and when the lifetime of CO<sub>2</sub> ranges from 30 to 95 years instead of 50 to 200 years. A similar curve, but based on a new set of



**Figure 3.** Corrected Figure 1 of J2002. The figure shows the comparative cooling of global climate due to eliminating all anthropogenic emissions of f.f. BC + OM, CH<sub>4</sub> (with a 10-year  $e$ -folding lifetime) and CO<sub>2</sub> (with 30-, 50-, and 95-year lifetimes). It is obtained from equation (7).

**Table 1.** Highest-Mileage Passenger Vehicles in the U.S. in 2005, Ranked by Their CO<sub>2</sub> Emissions (With and Without a Particle Trap in the Case of Diesel)<sup>a</sup>

Vehicle	Energy Source	Average Miles Per Gallon	CO <sub>2</sub> , g-C/km	CO <sub>2</sub> , g-C/km, With Trap
Honda Insight (M)	Gas/electric	63.5	23.4	
Honda Insight (A)	Gas/electric	56.5	26.2	
Toyota Prius (A)	Gas/electric	48.5	30.6	
Honda Civic (M)	Gas/electric	48.5	30.6	
Honda Civic (A)	Gas/electric	47.5	31.2	
Honda Civic (M)	Gas	40	37.1	
Toyota Echo (M)	Gas	38.5	38.5	
VW N. Beetle, Golf, Jetta (M)	Diesel	42	41.0	43.1
VW N. Beetle (A)	Diesel	39	44.1	46.3

<sup>a</sup>(A) denotes automatic transmission; (M) denotes manual transmission. The table assumes a gasoline and diesel density of 737 g/L and 840 g/L, respectively, a gasoline and diesel carbon content of 85.5% and 87.0%, respectively, and an increase in fuel use with a trap + filter of 5% (see text). Source of fuel economy is the Department of Energy (available at [www.fueleconomy.gov](http://www.fueleconomy.gov)).

simulations accounting for the effects of soot on snow albedo, is given by *Jacobson* [2004b].

[16] After the modification, Figure 3 still shows that controlling all f.f. BC + OM has an advantage over controlling all anthropogenic CO<sub>2</sub>, but for a shorter period (about 11–13 years) than does Figure 1 of J2002 (25–100 years). Thus the conclusion in J2002 that controlling f.f. BC + OM may be the most effective method of slowing global warming for a specific period still holds, but for a shorter period than originally estimated.

### 3. Comparison of Diesel Versus Gasoline

[17] Third, the comparison of diesel versus gasoline, embodied in Figure 14 of J2002, was updated to account for (1) the revision to Figure 1 of J2002, as shown in Figure 3 here and (2) a range of mileage differences of diesel versus gasoline rather than one difference. In addition, a lower estimate of the density of diesel (840 g/L) than the 856 g/L used in J2002, was assumed (a modification that benefits diesel).

[18] J2002 assumed that diesel vehicles obtained 30% better mileage than equivalent gasoline vehicles. This assumption, though, does not apply to the highest-mileage vehicles in the U.S. Table 1, for example, shows the highest-mileage diesel, gasoline, and gasoline-electric hybrid vehicle available in the U.S. in 2005. The table shows that the highest-mileage diesel vehicle obtains only 5% better mileage than does the highest-mileage gasoline vehicle (42 mpg versus 40 mpg). This translates into greater CO<sub>2</sub> emissions for the highest-mileage diesel vehicle since diesel fuel has a greater density and carbon content than does gasoline (Table 1). The addition of a particle trap to a diesel vehicle increases its fuel use by 3.5–8.5% [*Salvat et al.*, 2000; *Ullman et al.*, 2002; *Durbin and Norbeck*, 2002]. Assuming a 5% increase, diesels with a trap emit even more CO<sub>2</sub> per unit distance than do the gasoline vehicles (Table 1). In all cases, gasoline-electric hybrid vehicles available in the U.S. emit less CO<sub>2</sub> than do diesel with or without a trap and gasoline vehicles.

[19] Here, the effect of diesel versus gasoline on climate is reexamined when a range of mileage differences between diesel and gasoline (15–30% better for diesel instead of just 30% better, which was assumed in Figure 14 of J2002) is considered. When the mileage of a diesel is <13% better than that of gasoline (e.g., all cases in Table 1), gasoline vehicles

are always found to have a climate advantage, so no curves are shown for those cases. The updated result also accounts for the modified temperature-change curves in Figure 3 and a CO<sub>2</sub> lifetime range of 30–95 years.

[20] Figures 4a and 4b shows that when diesel vehicles achieve 30% or 15% higher mileage than do gasoline vehicles, diesel vehicles emitting particles continuously at a PM standard of 0.08 g/mi may warm climate more than gasoline vehicles for >100 years for all CO<sub>2</sub> lifetimes. When diesel achieves 15% higher, but not 30% higher, mileage than does gasoline, diesel vehicles emitting particles continuously at a tougher PM standard of 0.01 g/mi may also warm climate for more than 100 years.

[21] J2002, calculated that when diesel achieves 30% higher mileage than gasoline, diesel vehicles emitting 0.01 g/mi continuously for 100 years may warm climate for 13–54 years relative to gasoline vehicles. On the basis of the revised results in Figure 4b here, diesel may warm climate relative to gasoline for about 10 years at 30% higher mileage. Because no diesel vehicle available in the U.S. in 2005 emits less CO<sub>2</sub> than does the best gasoline vehicle available (Table 1), the 30% scenario is not applicable for the best available vehicles. As such, the upper end of the warming period due to diesel over gasoline must be >100 years.

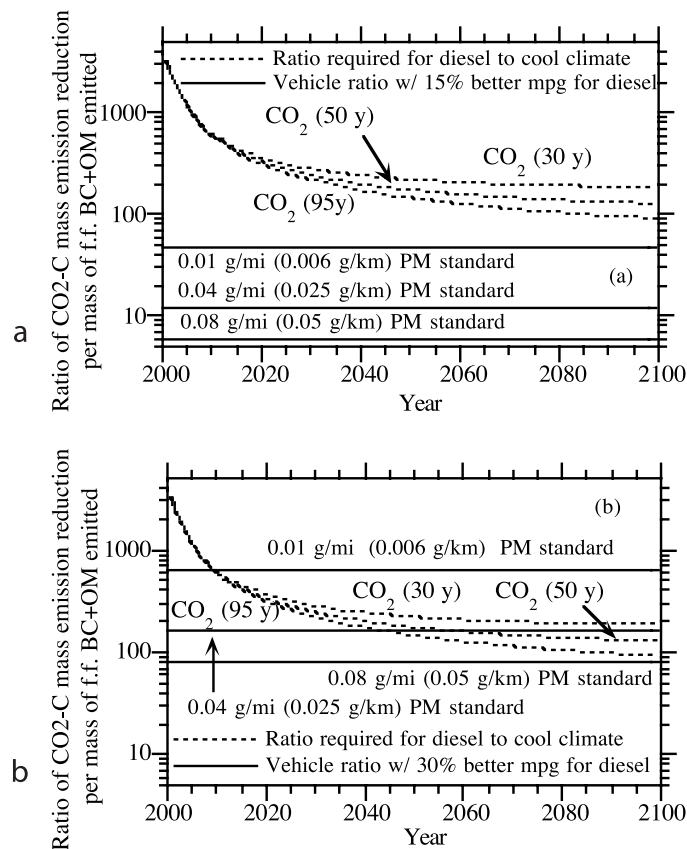
[22] Figure 4 (and Figure 14 of J2002) should be viewed cautiously, though, when considering the comparison at a 0.01 g/mi standard. First, regardless of whether gasoline or diesel cools at that level, the total mass of emission is small at that standard, so the magnitude of cooling or warming by either vehicle type at that level will be small. Second, gasoline vehicles also emit particles (generally 0.00008–0.003 g/mi, or 0.05–2 mg/km). Although such emissions are generally lower than those of diesel vehicles with a trap, Figure 4 can be applied correctly for the 0.01 g/mi standard only if it is assumed that diesel PM emissions are equal to gasoline PM emissions plus the standard.

[23] Finally, the caption from Figure 4 suggests that the 100-year climate-response per unit mass emission of f.f. BC + OM, relative to that of CO<sub>2</sub>-C, may range from about 90–190.

### 4. Summary

[24] Two figures in J2002 were updated. The updates do not change the main conclusions in J2002 regarding the relative benefit of f.f. BC + OM control versus CO<sub>2</sub> control





**Figure 4.** Comparison of the modeled ratio of the CO<sub>2</sub>-C emission reduction required per unit of f.f. BC + OM emitted for diesel vehicles to cool global climate with the actual ratio of CO<sub>2</sub>-C emission reduction per unit mass f.f. BC + OM emission when diesel achieves (a) 15% and (b) 30% better mileage than gasoline and when diesel has different f.f. BC + OM emission rates. The modeled curves (dashed lines) were obtained by dividing the f.f. BC + OM-temperature curve in Figure 3 by each CO<sub>2</sub>-temperature curve (30 years, 50 years, 95 years) then multiplying the result by the yearly emission rate of anthropogenic CO<sub>2</sub> (8100 Tg-C/yr) and dividing by that of BC and associated OM from fossil fuels (5.1 Tg/yr BC + 10.1 Tg/yr OM). The modeled curves show that a yearly 1 Tg/yr decrease in f.f. BC + OM emission cools climate by about 4200–4500 times more than does a 1 Tg/yr decrease in CO<sub>2</sub>-C emissions during 1 year. After 100 years of continuous 1 Tg/yr decreases in both, the resulting ratio of f.f. BC + OM to CO<sub>2</sub>-C cooling is about 90–190:1 (this ratio is the 100-year climate response of f.f. BC + OM per unit emission relative to that of CO<sub>2</sub>). The three solid, straight lines in each figure represent the actual ratio of CO<sub>2</sub>-C saved to f.f. BC + OM emitted for a modern diesel vehicle emitting 0.08, 0.04, and 0.01 g/mi BC + OM. The intersection of each straight line with each modeled curve indicates the period during which diesel vehicles enhance global warming in comparison with gasoline vehicles under the given emission standard. For example, in the case of the 0.08 g/mi standard, diesel warms climate in comparison with gasoline for >100 years for all CO<sub>2</sub> lifetimes and for both differences in diesel versus gasoline mileage.

and that of gasoline versus diesel, except that they modify the period over which f.f. BC + OM has an advantage.

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