# **Air Pollution and Global Warming:**

# History, Science, and Solutions

Cambridge University Press

# **Corrections and Modifications**

Mark Z. Jacobson
Department of Civil & Environmental Engineering

Last updated February 10, 2020

#### CHAPTER 1

- P. 25, Equation 1.13. Remove "+" after "N<sub>2</sub>O(g)"
- P. 27. Equation 1.14, The second "m" (in the numerator on the right side) should be in bold, nonitalics. Similarly, the "m" on the line below should be in bold, nonitalics.

#### **CHAPTER 2**

- P. 40. Second paragraph, change "oxygen levels increases" to "oxygen levels increased"
- P. 40. Second paragraph on left, change "Huronian glaciation (Section 12.3.2.1)" to "Huronian glaciation (Section 12.3.2.3)"

#### **CHAPTER 3**

P. 61. Add "of 390 ppmv" after "that gives a constant mixing ratio of CO<sub>2</sub>(g)" in the caption of Figure 3.13

#### **CHAPTER 4**

P. 99. Question 4.4. Change, "Why does the hydroxyl radical not form at night?" to "Why does the hydroxyl radical hardly form at night?"

# **CHAPTER 6**

P. 131. Figure 6.5. Change, "Supbolar" to "Subpolar" twice.

# **CHAPTER 7**

P. 167. Equation 7.11. The denominator should be  $I_B$  rather than I.

# **CHAPTER 13**

P. 339. Problem 13.4, last line. Multiply the equation by the efficiency rather than divide by the efficiency.

#### **APPENDIX**

P. 341. 1 bar =  $10^3$  hPa rather than  $10^5$  hPa.