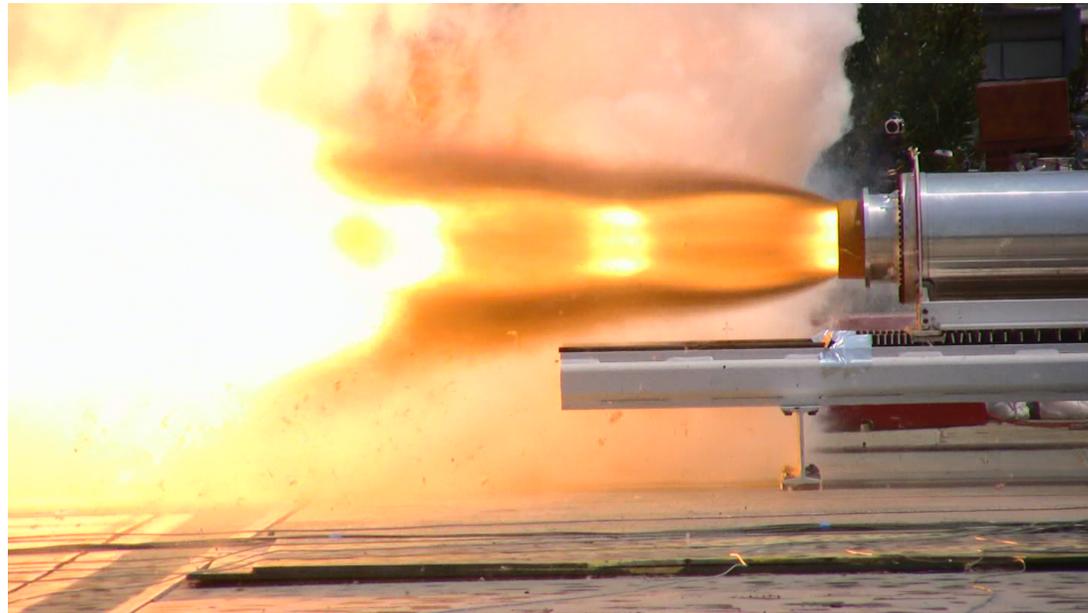


# Stanford Propulsion and Space Exploration Lab

Vision – Advance the science and technology of clean, low-cost propulsion and energy



<http://spase.stanford.edu>

<https://web.stanford.edu/~cantwell/>

Brian Cantwell  
Department of Aeronautics and Astronautics  
Stanford University

SPaSE lab Alumni at JPL working on space exploration



Beth Jens  
NASA – JPL

Perseverance cold gas  
subsystem for dust removal



Ashley Karp  
NASA – JPL

Propulsion Lead, Deputy Manager,  
Mars Ascent Vehicle (MAV)



Tim Szwarc  
NASA – JPL

Perseverance core  
drilling system



William Hart  
NASA – JPL

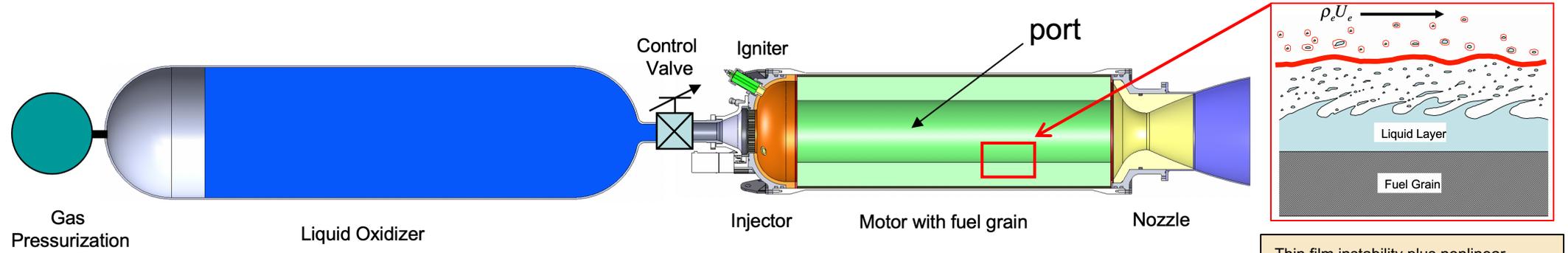
Cruise phase lead for mission Psyche



Flora Mechantel  
NASA – JPL

Project Starshade

# Liquifying fuels for hybrid propulsion - I



Thin film instability plus nonlinear growth and wave breaking leads to entrainment of droplets along the port

Karabeyoglu, Altman and Cantwell, "Combustion of liquifying hybrid propellants" parts 1 and 2, J. Prop. and Power, vol 18, 2002.

**15,000 lb thrust hybrid motor test**

**Fast burning paraffin-based fuels for hybrid rockets first identified in the SPaSE lab at Stanford by Arif Karabeyoglu. Large scale tests at NASA Ames led by Greg Zilliac.**

**Not all tests are succesful!**



**Visualization of hybrid rocket combustion, liquid layer instability and fuel droplet entrainment mechanism**

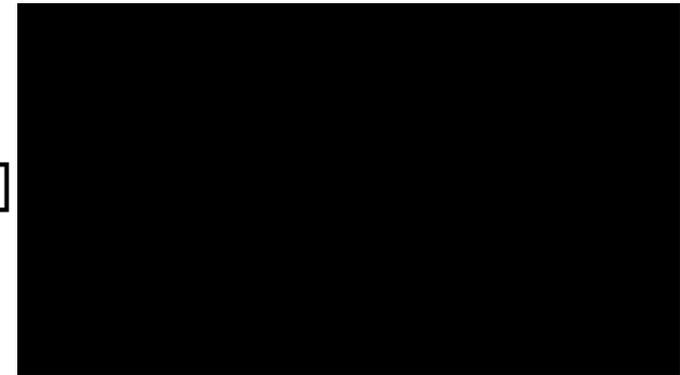
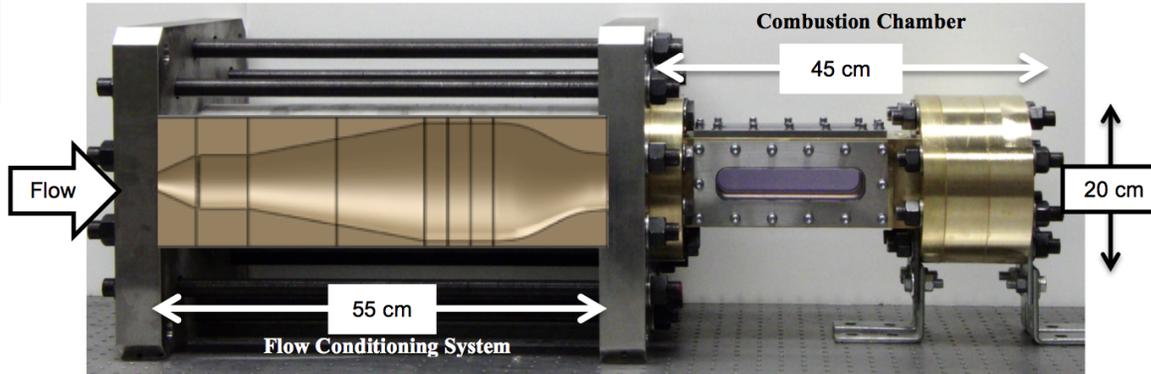
Oxygen compatible combustion visualization tunnel



Ashley Karp  
NASA - JPL



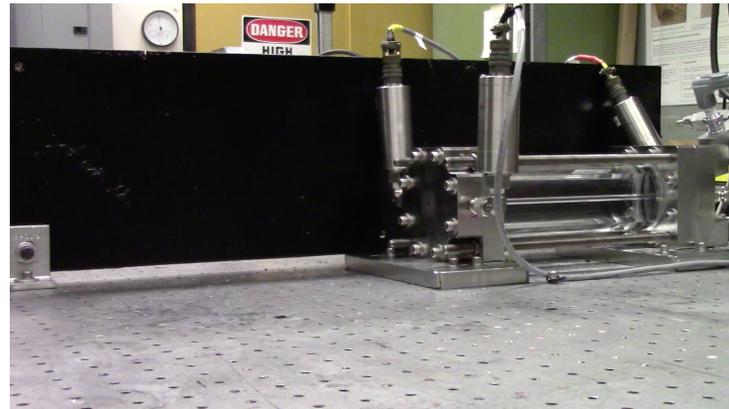
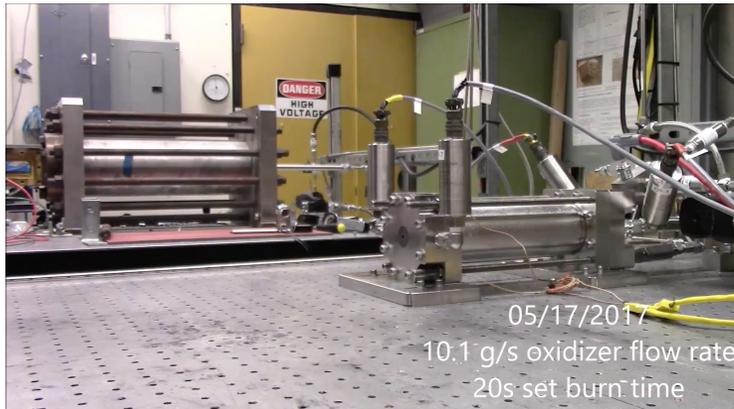
Beth Jens  
NASA - JPL



**Direct visualization of fuel regression in a clear-grain hybrid rocket**



Flora Mechantel  
NASA - JPL

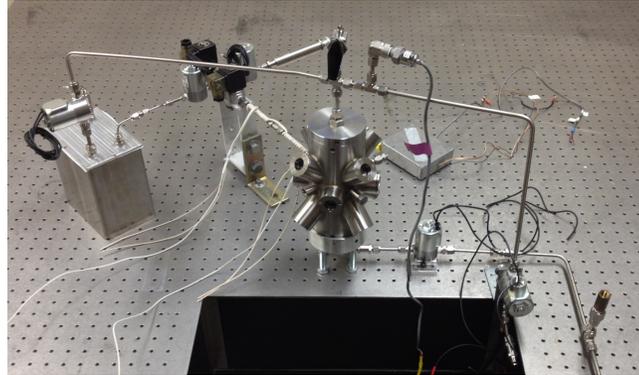


## Lithium aluminum hydride (LAH) and LAH doped paraffin reacting with Reagent grade Nitric acid

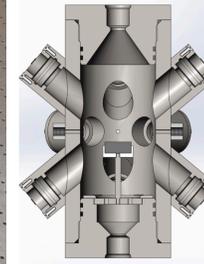


Javier Stober  
MIT

LAH reacting with 70% nitric acid at one atmosphere



Variable pressure droplet ignition chamber for studying ignition delay from vacuum to 3 MPa



Paraffin/40%LAH + 70% nitric acid at 95C



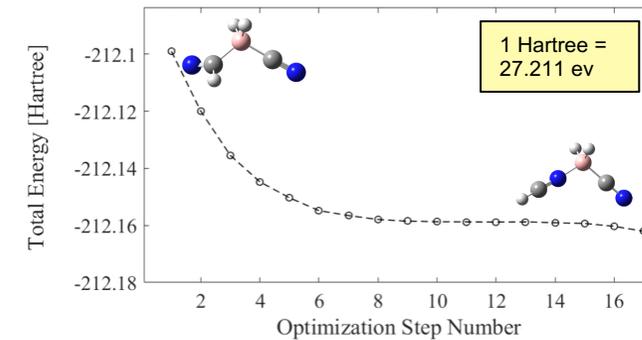
## Ignition and combustion of Energetic Ionic Liquids (EILs) reacting with Nitric Acid



Anna Thomas  
NREL



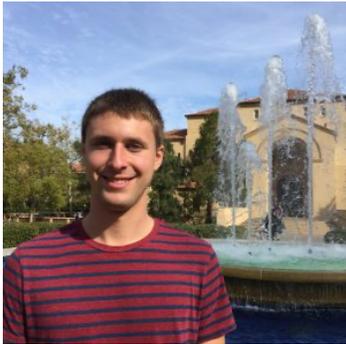
Figure 6.5: Stills taken from high speed video data recording the reaction between an ionic liquid, BMPY<sup>+</sup>DCA<sup>-</sup>, and WFNA at 3000 fps in a droplet ignition chamber providing a nitrogen inert environment. The droplet of the ionic liquid can be followed from the tip of the syringe needle until contact, mixing, and, finally, ignition with the WFNA.



Density Functional Theory (DFT) used to find the lowest energy configuration

- Decomposition reaction pathways were determined using
  - Thermogravimetric Analysis (TGA)
  - Fourier Transform Infrared Spectroscopy (FTIR)
  - Computational chemistry (code Gaussian)

Low power laser ignition of solid fuels for hybrid propulsion



David Dyrda  
Boeing Space Systems

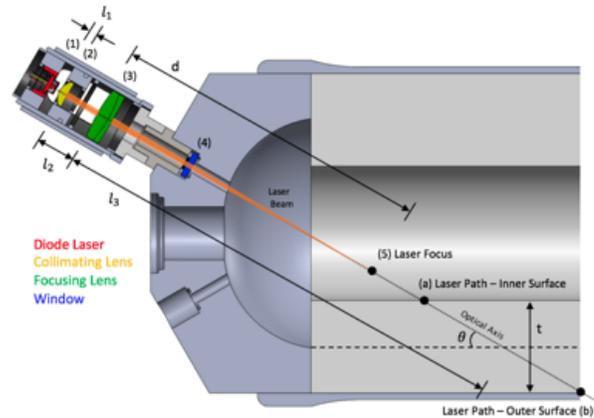


Figure 4: A representation of the proposed laser ignition concept for use in single port hybrid rocket motors. The laser system is composed of a single diode laser, a collimating and focusing lens, and a transparent window. The beam focuses to a point within the motor a distance  $d$  away from the focusing lens.

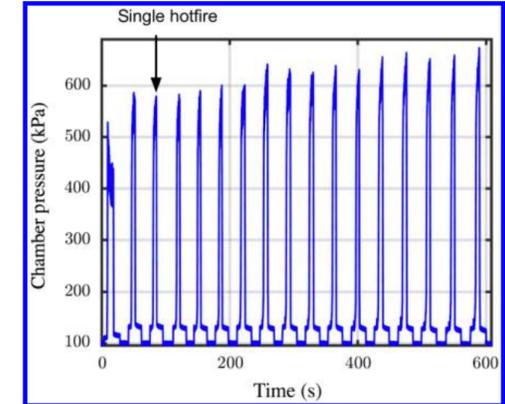
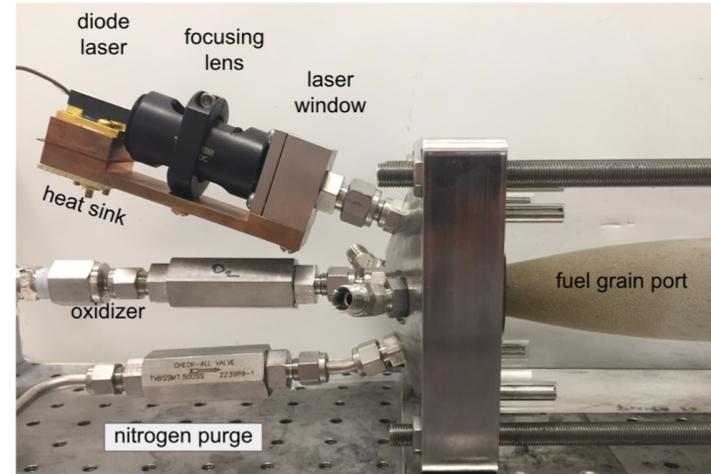


Fig. 4 Hot fires 1–17 on fuel grain 1, intended to mimic a series of restarts without motor adjustment. Chamber pressure data from individual tests have been stitched together.

Ignition mechanism – Soot particle entrainment and heating

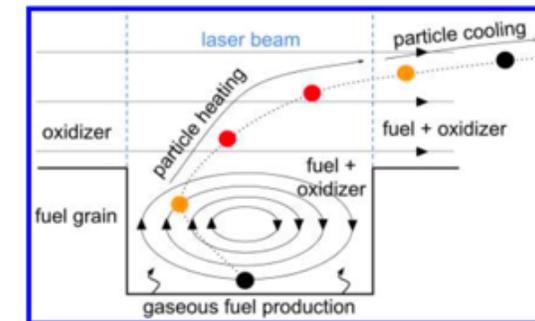
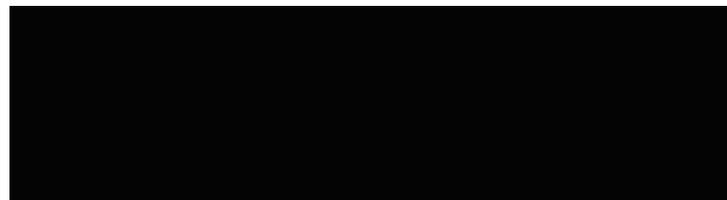
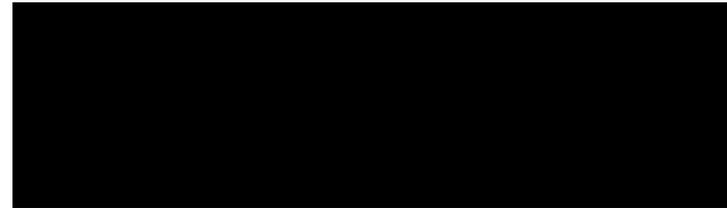
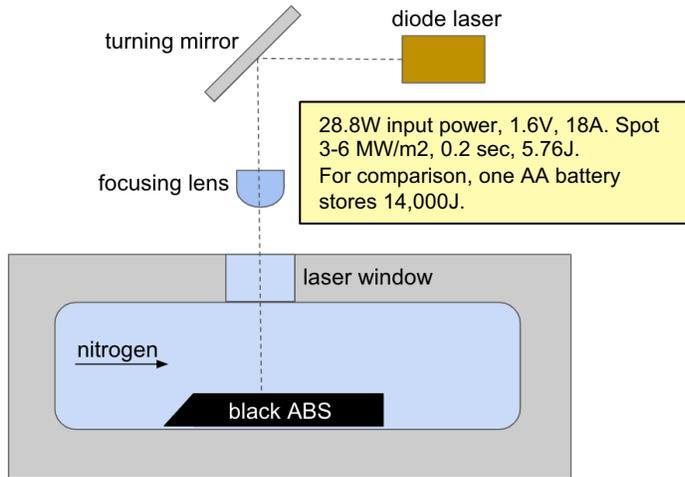
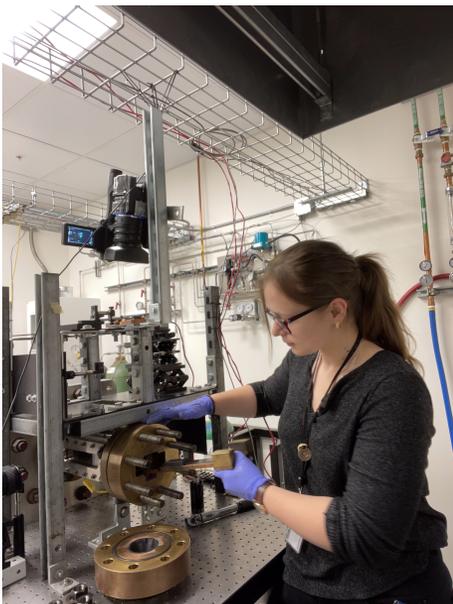
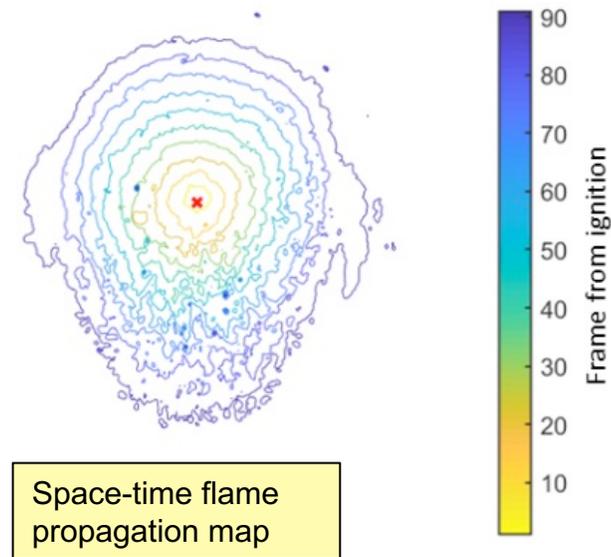
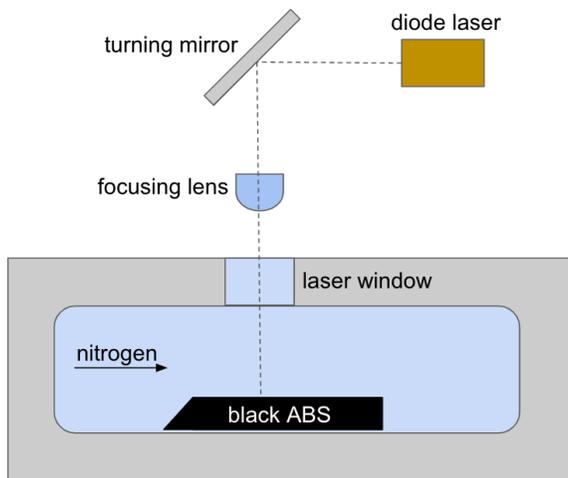
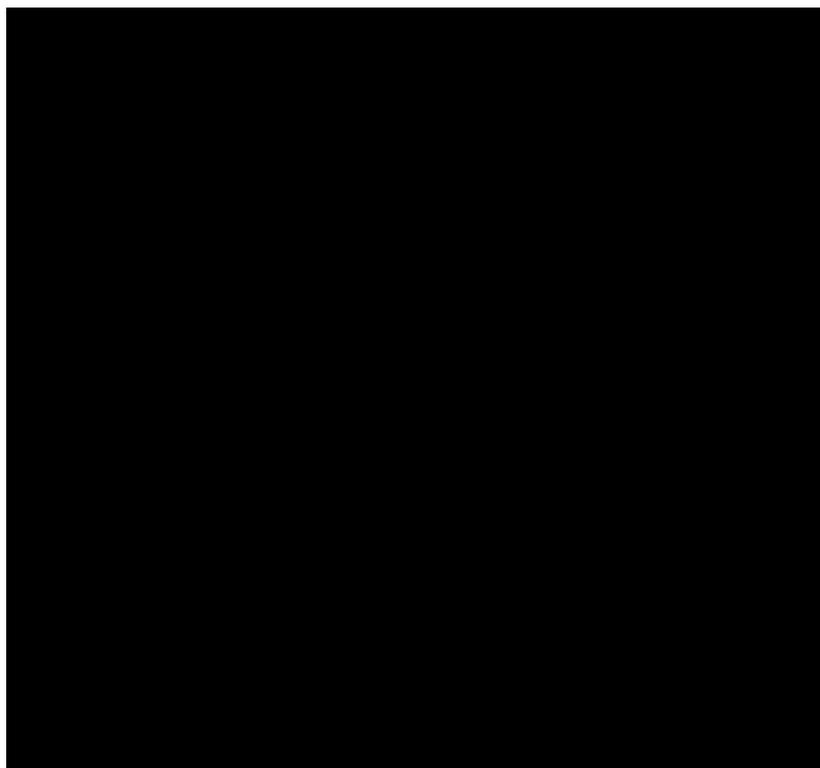
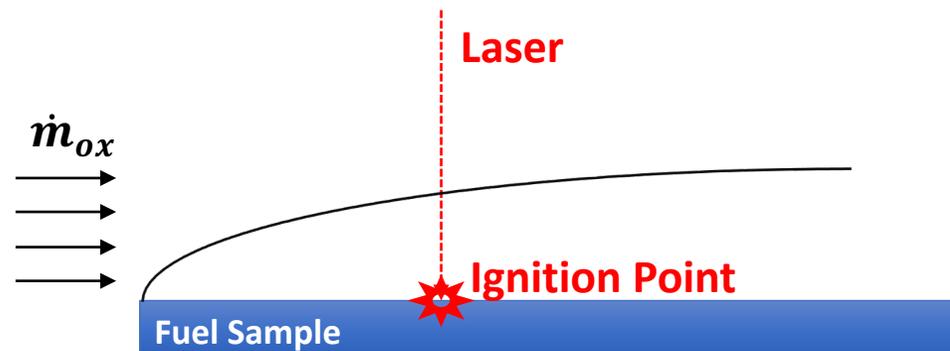


Fig. 5 Oxidizer flow over the laser cavity results in a recirculation zone that facilitates carbon particle entrainment. These particles are also heated by the laser.



Veronika Korneyeva  
Stanford AA

AIAA paper 2020 – 3731  
Experimental Studies of Flame Propagation on  
Planar Hybrid Fuel Samples in Opposed Flow  
AIAA Propulsion and Energy Forum Aug 2020



Space-time flame propagation map

# A Universal velocity profile for near-wall flows

## AIAA SciTech Conference Jan 2021

*J. Fluid Mech.* (2019), vol. 878, pp. 834–874. © The Author 2019  
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doi:10.1017/jfm.2019.669

834

### A universal velocity profile for smooth wall pipe flow

Brian J. Cantwell†

Department of Aeronautics and Astronautics, Stanford University, Stanford, CA 94305, USA

(Received 21 January 2019; revised 11 August 2019; accepted 11 August 2019)

AIAA 2021 - 0061

MATTHEW A. SUBRAHMANYAM, BRIAN J. CANTWELL, JUAN J. ALONSO

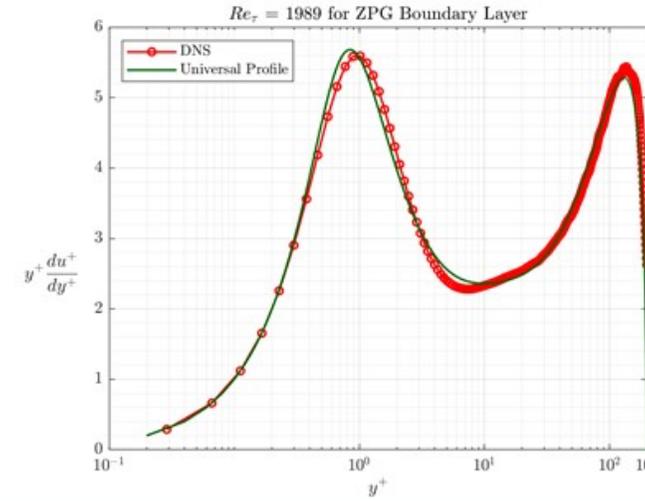
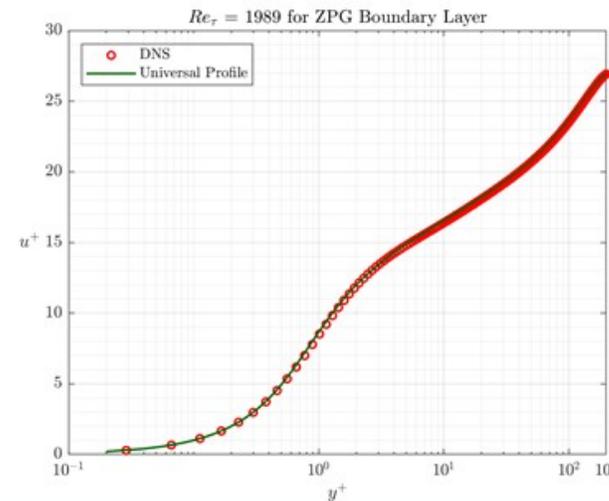
A NEW MIXING LENGTH MODEL HAS BEEN DEVELOPED TO ACCURATELY DESCRIBE WALL-BOUNDED FLOWS

Flow	a	b	k	m	n
Channel	22.8673	0.4649	0.4086	1.2569	1.3972
Pipe	20.0950	0.3195	0.4092	1.6210	1.6190
ZPG Boundary Layer	24.8462	0.1771	0.4193	1.2043	2.4331



Matt Subrahmanyam  
Stanford AA

A new turbulent boundary layer velocity profile is used that is uniformly valid from the wall to the free stream for all Reynolds numbers



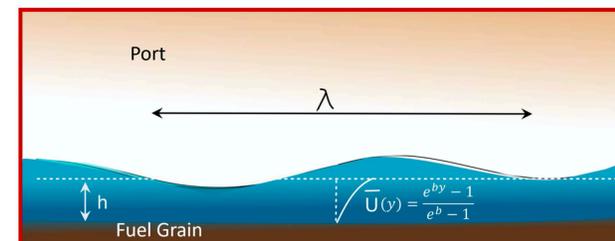
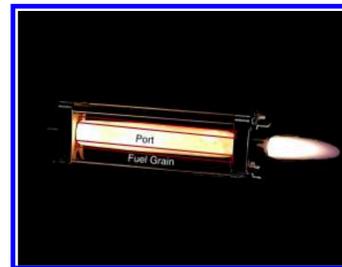
# Stability of thin liquid films subject to shear flow

## AIAA Propulsion and Energy Forum Conference Aug 2020



Eylul Bilgin  
Stanford AA

- AIAA-2020-3730
- Combustion of Liquefying Hybrid Propellants: Long-Wave Instabilities of Sheared Liquid Films Subject to Blowing
- Eylul Bilgin ([ebilgin@stanford.edu](mailto:ebilgin@stanford.edu)) Prof. Brian Cantwell ([cantwell@stanford.edu](mailto:cantwell@stanford.edu))
- **Summary**
  - Investigation of the long wave stability characteristics of pressure and shear driven liquid layers forming over liquefying hybrid propellants and the introduction of a mathematical technique for solving over-posed stability problems.
- **Key results**
  - **A new method (method of variational iterations) was used for the first time to solve a hydrodynamic stability problem.**
  - **The sensitivity of long wave stability characteristics to velocity curvature was demonstrated through both a numerical scheme and theoretical analysis.**



In collaboration with Prof. Craig Criddle of the Stanford Department of Civil and Environmental Engineering (CEE), This is the first project to propose a new process for recovering waste nitrogen from waste water for energy production.

## Energy & Environmental Science

RSC Publishing

PAPER

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### Nitrogen removal with energy recovery through N<sub>2</sub>O decomposition†

Cite this: *Energy Environ. Sci.*, 2013, 6, 241

Yaniv D. Scherson,<sup>a,e</sup> George F. Wells,<sup>b</sup> Sung-Geun Woo,<sup>d</sup> Jangho Lee,<sup>d</sup> Joonhong Park,<sup>d</sup> Brian J. Cantwell<sup>c</sup> and Craig S. Criddle<sup>b,e</sup>

#### Broader context

The release of reactive forms of nitrogen is a major environmental threat causing hypoxia and eutrophic zones in water bodies. Globally, rising energy costs and increasingly stringent discharge regulation are major drivers for efficient wastewater treatment processes that lower costs and increase recoverable energy from waste. While many processes recover energy from carbon waste as CH<sub>4</sub>, none recovers energy from waste nitrogen. This work introduces a new wastewater treatment process that removes and recovers energy from nitrogen waste by exploiting the thermodynamic properties of N<sub>2</sub>O for energy recovery. The proposed process, referred to here as Coupled Aerobic-anoxic Nitrous Decomposition Operation (CANDO), involves three steps: (1) partial aerobic nitrification of NH<sub>4</sub><sup>+</sup> to NO<sub>2</sub><sup>-</sup>, (2) partial anoxic denitrification of NO<sub>2</sub><sup>-</sup> to N<sub>2</sub>O, and (3) N<sub>2</sub>O conversion to N<sub>2</sub> with energy recovery *via* catalytic decomposition of N<sub>2</sub>O or use of N<sub>2</sub>O as an oxidant in CH<sub>4</sub> combustion. If successfully scaled-up, this process has the potential to lower aeration and biosolid production (the two major operational costs), increase CH<sub>4</sub> recovery from “freed” organic matter, and introduces a new renewable energy source from CH<sub>4</sub> combustion with N<sub>2</sub>O.



Prof. Craig Criddle



Yaniv Scherson  
Anaergia

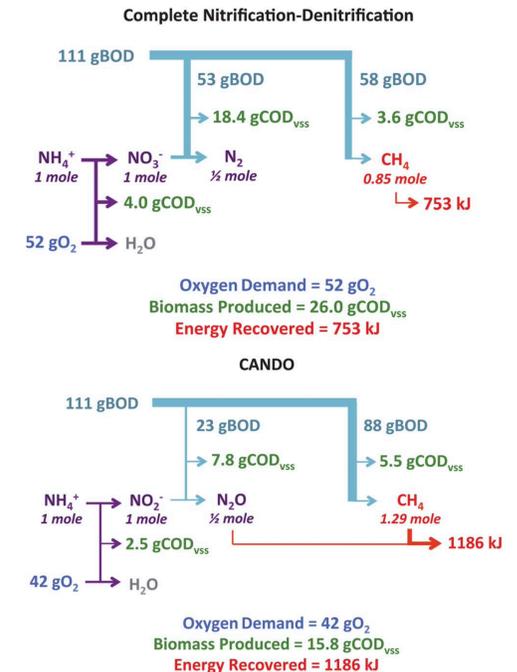
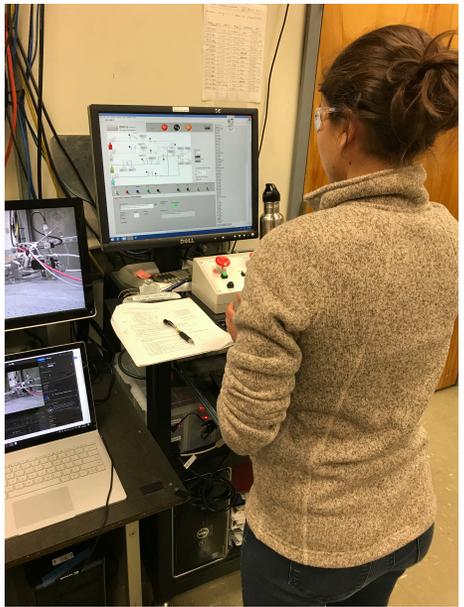
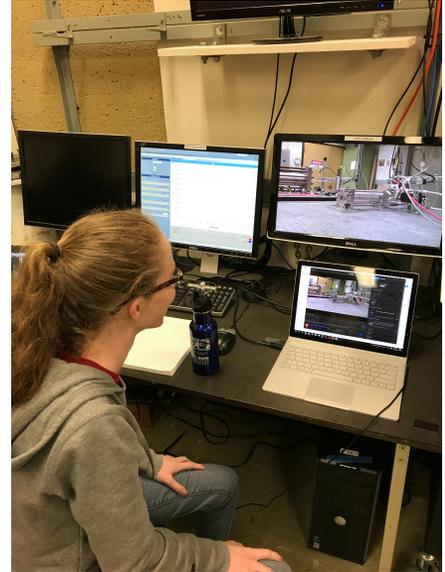
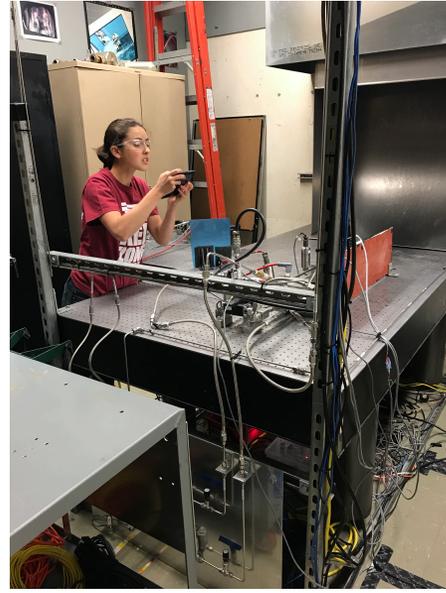
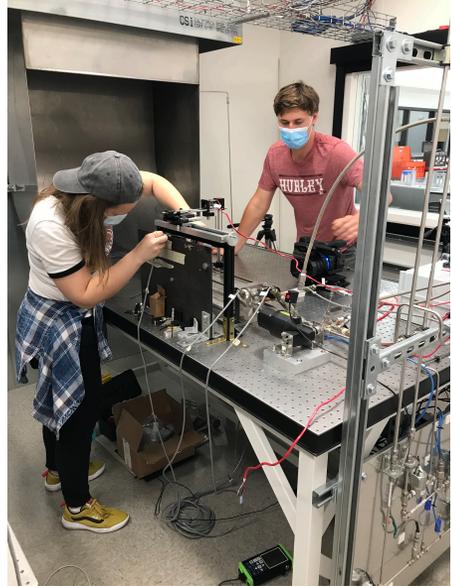
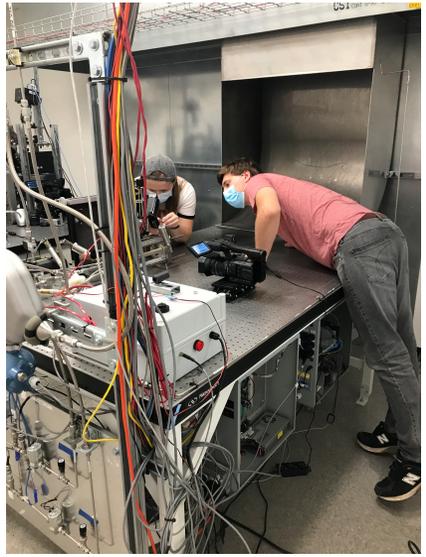


Fig. 8 Comparison of complete nitrification–denitrification to CANDO with respect to oxygen demand, biomass production, and energy recovery.

# Working in the SPaSE lab



3/2/21

## Recent PhD grads



Jonah Zimmerman  
Blue Origin



Ashley Karp  
NASA - JPL



Tim Szwarc  
NASA - JPL



Ashley Micks  
Ford Motor Co



Laura Simurda  
Stanford AA



Beth Jens  
NASA - JPL



Yaniv Scherson  
Anaergia



Javier Stober  
MIT



Flora Mechtel  
NASA - JPL



Ashley Coates  
NASA Ames



David Murakami  
NASA Ames



Pavan Narsai  
Momentus Corp.



Kevin Lohner  
SpaceX



David Dyrda  
Boeing Space Systems



Ben Waxman  
Blue Origin  
3/2/21

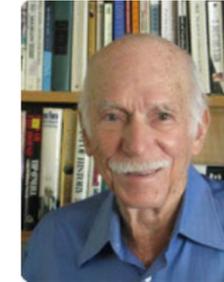


Anna Thomas  
NREL

## Colleagues



Brian Cantwell, Arif Karabeyoglu, Greg Ziliac



Dave Altman  
Vice pres CSD,  
Ret.



Scott Hubbard  
Stanford

## Current PhD students



Matt  
Subrahmanyam  
Stanford AA



Eylul Bilgin  
Stanford AA



Veronika Korneyeva  
Stanford AA